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(54) Title: VAPOR PHASE SILOXANE DRY CLEANING PROCESS

(57) Abstract: The process of the present invetion is direced to a dry cleaning process, comprising the use of volatile cyclic, linear or branched siloxanes in the vapor phase for the cleaning of soiled or staned fabrics. The linear or branched siloxanes have the formula: $M_{2+p+2z}D_xT_yQ_z$ wherein: M is $R_3^1SiO_{1/2}$; D is $R^2R_3^3SiO_{2/2}$; T is $R^4SiO_{3/2}$; and Q is $SIO_{4/2}$. R^1 , R^2 , R^3 and R^4 are each independently a monovalent hydrocarbaon radical having form one to forthy carbon stoms; and x and y are each integers, wherein 0 < x < 10 and 0 < y < 10. While the cyclic siloxanes have the formula (1) wherein R^5 , R^6 , R^7 and R^8 are each independently a monovalent hydrocarbon group having from one to forty carbon atoms; and a and b are each integers wherein 0 < a < 10 and 0 < b < 10, provided that 3 < (a + b) < 10.

02/07/356

VAPOR PHASE SILOXANE DRY CLEANING PROCESS

TECHNICAL FIELD

The present invention is directed to a dry cleaning process, more specifically, to a siloxane vapor phase based process, for use in dry cleaning.

BACKGROUND

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Current dry cleaning technology uses perchloroethylene ("PERC") or petroleum-based materials as the cleaning solvent. PERC suffers from toxicity and odor issues. The petroleum-based products are not as effective as PERC in cleaning garments.

Cyclic siloxanes have been reported as spot cleaning solutions, see US 4,685,930, and as dry cleaning fluids in dry cleaning machines, see US 5,942,007. Other patents disclose the use of silicone soaps in petroleum solvents, see JP 09299687, and the use of silicone surfactants in super critical carbon dioxide solutions has been reported, see, for example, US 5,676,705 and Chem. Mark. Rep., 15 Dec 1997, 252(24), p. 15. Non-volatile silicone oils have also been used as the cleaning solvent requiring removal by a second washing with perfluoroalkane to remove the silicone oil, see JP 06327888.

Numerous other patents have issued in which siloxanes or organomodified silicones have been present as addenda in PERC or petroleum based dry cleaning solvents, see, for example, WO 9401510; US 4911853; US 4005231; US 4065258.

There is a continued interest in decreasing the need for large quantities of solvents used in dry cleaning processes.

SUMMARY OF THE INVENTION

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The process of the present invention is directed to a cleaning process, comprising the use of a volatile cyclic, linear or branched siloxane in the vapor phase for the cleaning of articles.

Further, the present invention provides for a process for cleaning soiled articles of manufacture comprising:

- a) contacting the soiled article of manufacture with a vapor phase silicone compound;
- allowing the vapor phase silicone compound in contact with the soiled article of manufacture to condense to the liquid phase becoming thereby a condensed silicone liquid; and
- c) draining the condensed silicone liquid away from the article of manufacture whereby the soiled article of manufacture is cleaned.

In another embodiment the present invention provides for a process for cleaning soiled garments comprising:

- a) contacting the soiled garment with a vapor phase silicone compound;
 - allowing the vapor phase silicone compound in contact with the soiled garment to condense to the liquid phase becoming thereby a condensed silicone liquid; and
- c) draining the condensed silicone liquid away from the garment whereby
 the soiled garment is cleaned.

DETAILED DESCRIPTION OF THE INVENTION

The compounds useful in the practice of the present invention may be linear, branched or cyclic volatile siloxane compounds. In general those siloxanes that are volatile and suitable for use in the practice of the present invention are those siloxanes that are volatile at room temperature, i.e. about 25 °C. Volatility is a quantitative measurement at a given temperature and thus broadly defined involves a partial pressure or vapor pressure, i.e. a pressure below 760 mm Hg, at a given temperature. Broadly, volatile siloxanes are those siloxanes that have a vapor pressure or partial pressure (as used herein the two terms are interchangeable) above 0.01 mm Hg at a temperature of 20 °C.

Compounds suitable as the linear or branched, volatile siloxane solvent of the present invention are those containing a polysiloxane structure that includes from 2 to 20 silicon atoms. Preferably, the linear or branched, volatile siloxanes are relatively volatile materials, having, for example, a boiling of below about 300°C point at a pressure of 760 millimeters of mercury ("mm Hg").

In one embodiment, the linear or branched, volatile siloxane comprises one or more compounds of the structural formula (I):

 $M_{2+y+2z}D_xT_yQ_z$ (I)

20 wherein:

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M is R¹₃SiO_{1/2};

D is R²R³SiO_{2/2};

T is $R^4SiO_{3/2}$;

and Q is SiO_{4/2}

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R¹, R², R³ and R⁴ are each independently a monovalent hydrocarbon radical having from one to forty carbon at0ms; and

x and y are each integers, wherein $0 \le x \le 10$ and $0 \le y \le 10$ and $0 \le z \le 10$.

Suitable monovalent hydrocarbon groups include linear hydrocarbon radicals, branched hydrocarbon radicals, monovalent alicyclic hydrocarbon radicals, monovalent and aromatic or fluoro containing hydrocarbon radicals. Preferred monovalent hydrocarbon radicals are monovalent alkyl radicals, monovalent aryl radicals and monovalent aralkyl radicals.

As used herein, the term "(C₁-C₆)alkyl" means a linear or branched alkyl group containing from 1 to 6 carbons per group, such as, for example, methyl, ethyl, propyl, iso-propyl, n-butyl, iso-butyl, sec-butyl, tert-butyl, pentyl, hexyl, preferably methyl.

As used herein, the term "aryl" means a monovalent unsaturated hydrocarbon ring system containing one or more aromatic or fluoro containing rings per group, which may optionally be substituted on the one or more aromatic or fluoro containing rings, preferably with one or more (C1-C6) alkyl groups and which, in the case of two or more rings, may be fused rings, including, for example, phenyl, 2,4,6-trimethylphenyl, 2-isopropylmethylphenyl, 1-pentalenyl, naphthyl, anthryl, preferably phenyl.

As used herein, the term "aralkyl" means an aryl derivative of an alkyl group, preferably a (C₂-C₆)alkyl group, wherein the alkyl portion of the aryl derivative may, optionally, be interrupted by an oxygen atom, such as, for example, phenylethyl, phenylpropyl, 2-(1-naphthyl)ethyl, preferably phenylpropyl, phenyoxypropyl, biphenyloxypropyl.

In another embodiment, the monovalent hydrocarbon radical is a monovalent (C_1 - C_6)alkyl radical, most preferably, methyl.

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In another embodiment, the linear or branched, volatile siloxane comprises one or more of, hexamethyldisiloxane, octamethyltrisiloxane, decamethyltetrasiloxane, dodecamethylpentasiloxane, tetradecamethylhexasiloxane or hexadecamethylheptasiloxane or methyltris(trimethylsiloxy)silane. In a more highly preferred embodiment, the linear or branched, volatile siloxane of the present invention comprises octamethyltrisiloxane, decamethyltetrasiloxane, or dodecamethylpentasiloxane or methyltris(trimethylsiloxy)silane. In a highly preferred embodiment, the siloxane component of the composition of the present invention consists essentially of decamethyltetrasiloxane.

Suitable linear or branched volatile siloxanes are made by known methods, such as, for example, hydrolysis and condensation of one or more of tetrachlorosilane, methyltrichlorosilane, dimethyldichlorosilane, trimethylchlorosilane, or by isolation of the desired fraction of an equilibrate mixture of hexamethyldisiloxane and octamethylcyclotetrasiloxane or the like and are commercially available.

Compounds suitable as the cyclic siloxane component of the present invention are those containing an oligomeric or polysiloxane ring structure that includes from 2 to 20 silicon atoms in the ring. Preferably, the linear, branched and cyclic siloxanes are relatively volatile materials, having, for example, a boiling point of below about 300°C at a pressure of 760 millimeters of mercury ("mm Hg"). Thus for the purposes of defining a volatile siloxane compound useful in the practice of the process of the present invention a volatile siloxane, whether linear branched or cyclic has a vapor pressure

ranging from 0.01 to 760 mm Hg at a temperature ranging from about 10 ℃ to about 300 ℃.

In another embodiment, the cyclic siloxane comprises one or more compounds of the structural formula (II):

(II)

wherein:

R⁵, R⁶, R⁷ and R⁸ are each independently a monovalent hydrocarbon group having from one to forty carbon atoms; and

a and b are each integers wherein $0 \le a \le 10$ and $0 \le b \le 10$, provided that $3 \le (a + b) \le 10$.

In yet another embodiment, the cyclic siloxane comprises one or more of, octamethylcyclotetrasiloxane, decamethylcyclopentasiloxane, dodecamethylcyclohexasiloxane, tetradecamethylcycloheptasiloxane. In a more highly preferred embodiment, the cyclic siloxane of the present invention comprises octamethylcyclotetrasiloxane or decamethylcyclopentasiloxane. In yet another embodiment, the cyclic siloxane component of the composition of the present invention consists essentially of decamethylcyclopentasiloxane.

Suitable cyclic siloxanes are made by known methods, such as, for example, hydrolysis and condensation of alkylhalosilanes, e.g. dimethyldichlorosilane, and are commercially available.

The process of the invention involves generation of a gas phase silicone by a combination of heating the silicone in a solvent reservoir and optionally reducing pressure so as to allow the silicone to vaporize followed by contacting the silicone vapors with the garment to be cleaned. While the formula describing compounds useful in the process of the present invention has already been described, those compounds useful in the practice of the present invention in one embodiment should have vapor pressures between about 3.0 mm Hg and about 760 mm Hg at temperatures ranging from about 20 °C to about 100 °C. In a second embodiment those compounds useful in the practice of the present invention should have vapor pressures between about 0.01 mm Hg and about 760 mm Hg at temperatures ranging from about 20 ℃ to about 270 ℃. In a third embodiment those compounds useful in the practice of the present invention should have vapor pressures between about 1.0 mm Hg and about 760 mm Hg at temperatures ranging from about 20 °C to about 134 °C. In a fourth embodiment those compounds useful in the practice of the present invention should have vapor pressures between about 0.01 mm Hg and about 760 mm Hg at temperatures ranging from about 20 ℃ to about 264 ℃.

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Alternatively, other methods known in the art can be employed to form vapors of silicones including mechanical means.

The vapors of the compounds of the present invention thus formed, either at atmospheric pressure or at reduced pressure, are allowed to contact the fabric to be cleaned for a specified time wherein these same vapors condense in the fabric, dissolving the soiling material or stain and draining away from the fabric, after which time the articles are removed, cooled as needed, and dried by various methods known in the art such as air drying, heated drying and the like. In one embodiment, the process of the present invention may be performed at a constant pressure. In another embodiment

the process of the present invention is performed at a pressure that is varied among the steps of the process, e.g. initially contacting the garment to be cleaned with a vapor at a pressure below atmospheric followed by raising the pressure to atmospheric pressure to condense the vapor in the garment and allow the cleaning fluids to drain away from the garment.

Alternatively, the articles remain in the cleaning vessel and the silicone or silicone containing solvent is removed by various means and the articles are dried in the cleaning vessel as is commonly seen in typical dry cleaning machines.

An article, such as for example, a textile or leather article, typically, a garment, is cleaned by contacting the article with the vapors of the composition of the present invention. In a preferred embodiment, the articles to be cleaned include textiles made from natural fibers, such as for example, cotton, wool, linen and hemp, from synthetic fibers, such as, for example, polyester fibers, polyamide fibers, polypropylene fibers and elastomeric fibers, from blends of natural and synthetic fibers, from natural or synthetic leather or natural or synthetic fur.

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The article and dry cleaning composition are then separated, by, for example, one or more of draining and centrifugation. In a preferred embodiment, separation of the article and dry cleaning composition is followed by the application of heat, preferably, heating to a temperature of from 15°C to 120°C, preferably from 20°C to 100°C, or reduced pressure, preferably, a pressure of from 1 mm Hg to 750 mm Hg, or by application of both heat and reduced pressure, to the article.

Testing for oil soluble stain removal was accomplished using a blue 50/50 cotton/poly cloth and a red satin fabric. The approximately 2 inch square samples were stained with motor oil, suspended by wires in a large

glass vessel equipped with a thermometer, and condensing unit capable of condensing the volatile silicone solvent. The articles were positioned such that the solvent vapors saturated the article but were not contacted by the returning, condensed solvent.

The process of the present invention is not limited to the cleaning of garments or articles of clothing, it may be applied to any article of manufacture contaminated with a silicone soluble contaminant that may be subjected to the process of the present invention wherein the contaminant is dissolved in the silicone compound and drained away, thereby removing the contaminant from the article of manufacture.

The following examples are to illustrate the invention and are not to be construed as limiting the claims.

EXAMPLES

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Example 1 - Atmospheric pressure, cyclic solvent

Samples of red satin and blue cotton/poly fabrics were treated with motor oil which was allowed to stain for 18 hours then attached to a wire holder and suspended above a reservoir of D5. The solvent was heated to boiling and the vapors allowed to contact the stained fabrics for 5 minutes. After this time, the heat was removed, the vessel cooled and the samples removed and air dried and evaluated. All traces of the oil were removed from both fabrics. There was some extraction of the red dye from the satin fabric.

Example 2 - Reduced Pressure, cyclic solvent

Samples of red satin and blue cotton/poly fabrics were treated with motor oil which was allowed to stain for 18 hours then attached to a wire holder and suspended above a reservoir of D5. The pressure in the system

was reduced to 1-2 mm Hg and the temperature of the solvent reservoir was raised to 70-80 °C. The vapors were allowed to contact the stained fabrics for 5 minutes. After this time, the heat was removed, the vessel cooled and the samples removed and air dried and evaluated. All traces of the oil were removed from both fabrics. No extraction of the red dye from the satin fabric was observed.

Example 3 - Reduced Pressure, linear solvent

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Samples of red satin and blue cotton/poly fabrics were treated with motor oil which was allowed to stain for 18 hours then attached to a wire holder and suspended above a reservoir of MD2M. The pressure in the system was reduced to 1-2 mm Hg and the temperature of the solvent reservoir was raised to 70-80 °C. The vapors were allowed to contact the stained fabrics for 5 minutes. After this time, the heat was removed, the vessel cooled and the samples removed and air dried and evaluated. All traces of the oil were removed from both fabrics. No extraction of the red dye from the satin fabric was observed.

Having described the invention, that which is claimed is:

- A process for cleaning soiled garments comprising:
 - a) contacting the soiled garment with a vapor phase silicone compound;
 - allowing the vapor phase silicone compound in contact with the soiled garment to condense to the liquid phase becoming thereby a condensed silicone liquid; and
 - draining the condensed silicone liquid away from the garment whereby the soiled garment is cleaned.
- The process of claim 1 wherein the silicone compound has the formula:

 $M_{2+y+2z}D_{x}T_{y}Q_{z}$

wherein:

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M is R13SiO1/2;

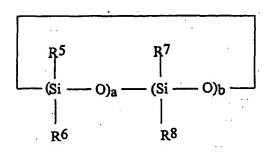
D is $R^2R^3SiO_{2/2}$;

T is $R^4SiO_{3/2}$;

15 and Q is SiO_{4/2}

 R^1 , R^2 , R^3 and R^4 are each independently a monovalent hydrocarbon radical having from one to forty carbon atoms; and x and y are each integers, wherein $0 \le x \le 10$ and $0 \le y \le 10$ and $0 \le z \le 10$.

3. The process of claim 1 wherein the silicone compound has the formula:



wherein:

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R⁵, R⁶, R⁷ and R⁸ are each independently a monovalent hydrocarbon group having from one to forty carbon atoms; and

a and b are each integers wherein $0 \le a \le 10$ and $0 \le b \le 10$, provided that $3 \le (a + b) \le 10$.

- 4. The process of claim 2 wherein each of the steps a), b) and c) are independently conducted at a temperature ranging from about 10 ℃ to about 300 ℃.
- 5. The process of claim 3 wherein each of the steps a), b) and c) are independently conducted at a temperature ranging from about 10 °C to about 300 °C.
- 6. The process of claim 4 wherein each of the steps a), b) and c) are independently conducted at a pressure ranging from about 0.01 mm Hg to about 760 mm Hg.
 - 7. The process of claim 5 wherein each of the steps a), b) and c) are independently conducted at a pressure ranging from about 0.01 mm Hg to about 760 mm Hg.

8. The process of claim 6 wherein the silicone compound is selected from the group consisting of hexamethyldisiloxane, octamethyltrisiloxane, decamethyltetrasiloxane, dodecamethylpentasiloxane, tetradecamethylhexasiloxane, hexadecamethylheptasiloxane and methyltris(trimethylsiloxy)silane.

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- 9. The process of claim 7 wherein the silicone compound is selected from the group consisting of hexamethyldisiloxane, octamethyltrisiloxane, decamethyltetrasiloxane, dodecamethylpentasiloxane, tetradecamethylhexasiloxane, hexadecamethylheptasiloxane and methyltris(trimethylsiloxy)silane.
- 10. A process for cleaning soiled articles of manufacture comprising:
 - a) contacting the soiled article of manufacture with a vapor phase silicone compound;
- allowing the vapor phase silicone compound in contact with the soiled article of manufacture to condense to the liquid phase becoming thereby a condensed silicone liquid; and
 - c) draining the condensed silicone liquid away from the article of manufacture whereby the soiled article of manufacture is cleaned.
- 11. A process for cleaning soiled garments consisting essentially of:
- a) contacting the soiled garment with a vapor phase silicone compound;
 - allowing the vapor phase silicone compound in contact with the soiled garment to condense to the liquid phase becoming thereby a condensed silicone liquid; and
 - c) draining the condensed silicone liquid away from the garment whereby the soiled garment is cleaned.

12. The process of claim 11 wherein the silicone compound has the formula:

 $M_{2+y+2z}D_xT_yQ_z$

wherein:

M is R¹3SiO_{1/2};

D is R2R3SiO2/2;

T is R4SiO3/2;

and Q is SiO_{4/2}.

 R^1 , R^2 , R^3 and R^4 are each independently a monovalent hydrocarbon radical having from one to forty carbon at0ms; and x and y are each integers, wherein $0 \le x \le 10$ and $0 \le y \le 10$ and $0 \le z \le 10$.

13. The process of claim 11 wherein the silicone compound has the formula:

15 wherein:

R⁵, R⁶, R⁷ and R⁸ are each independently a monovalent hydrocarbon group having from one to forty carbon atoms; and a and b are each integers wherein $0 \le a \le 10$ and $0 \le b \le 10$, provided that $3 \le (a + b) \le 10$.

14. The process of claim 12 wherein each of the steps a), b) and c) are independently conducted at a temperature ranging from about 10 °C to about 300 °C.

- 15. The process of claim 13 wherein each of the steps a), b) and c) are independently conducted at a temperature ranging from about 10 ℃ to about 300 ℃.
 - 16. The process of claim 14 wherein each of the steps a), b) and c) are independently conducted at a pressure ranging from about 0.01 mm Hg to about 760 mm Hg.
- 17. The process of claim 15 wherein each of the steps a), b) and c) are independently conducted at a pressure ranging from about 0.01 mm Hg to about 760 mm Hg.
- 18. The process of claim 16 wherein the silicone compound is selected from the group consisting of hexamethyldisiloxane, octamethyltrisiloxane, decamethyltetrasiloxane, dodecamethylpentasiloxane, tetradecamethylhexasiloxane, hexadecamethylheptasiloxane and methyltris(trimethylsiloxy)silane.
- 19. The process of claim 17 wherein the silicone compound is selected from the group consisting of hexamethyldisiloxane, octamethyltrisiloxane, decamethyltetrasiloxane, dodecamethylpentasiloxane, tetradecamethylhexasiloxane, hexadecamethylheptasiloxane and methyltris(trimethylsiloxy)silane.

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A. CLASSIFICATION OF SUBJECT MATTER
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According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols) IPC 7 D06L C11D D06F

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the International search (name of data base and, where practical, search terms used)

EPO-Internal, WPI Data

C DOCUME	ENTS CONSIDERED TO BE RELEVANT	•
Category *	Citation of document, with indication, where appropriate, of the relevant	it passages Relevant to claim No.
X	DATABASE WPI Section Ch, Week 197826 Derwent Publications Ltd., London, 6 Class A26, AN 1978-46701A XP002206086	1,2,4,6, 8,10-12, 14,16,18
	& JP 53 056203 A (LION FAT & OIL CO 22 May 1978 (1978-05-22) abstract) LTD),
X	US 5 834 416 A (MORGAN DAVID LEE E 10 November 1998 (1998-11-10) column 6, line 44 - line 53 claims 1,2	T AL) 10-12, 14,16,18
X	WO 99 10587 A (LAUBACH BERNADETTE S; MICELL TECHNOLOGIES (US); DEYOUNG 4 March 1999 (1999-03-04) page 8, line 1 - line 32	STOREY 1,10,11 JAMES)
X Fur	her documents are listed in the continuation of box C.	Patent family members are listed in annex.
"A" docum consk "E" earlier filing o "L" docum which citatio "O" docum other	ent defining the general state of the art which is not lered to be of particular relevance document but published on or after the international late with which may throw doubts on priority claim(s) or is cited to establish the publication date of another or or other special reason (as specified) ent referring to an oral disclosure, use, exhibition or means ent published prior to the international filing date but	later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken atone document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.
	actual completion of the international search	Date of mailing of the international search report
	6 July 2002 mailing address of the ISA European Patent Office, P.B. 5818 Patentiaan 2 NI. – 2280 HV Rijswijk Tel. (+31-70) 340-2040, Tx. 31 651 epo ni, Fax: (+31-70) 340-3016	29/07/2002 Authorized officer Richards, M

in ional Application No

C.(Contin	uation) DOCUMENTS CONSIDERED TO BE RELEVANT	PC1/US 02/04620
Category °	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to ctalm No.
P,X	EP 1 092 803 A (UNILEVER PLC ;UNILEVER NV (NL)) 18 April 2001 (2001-04-18) example 3	1,10,11
Х	DATABASE WPI Section Ch, Week 199439 Derwent Publications Ltd., London, GB; Class A97, AN 1994-312983 XP002206053 & JP 06 238244 A (JAPAN FIELD KK), 30 August 1994 (1994-08-30) abstract	10,11
X	DATABASE WPI Section Ch, Week 199608 Derwent Publications Ltd., London, GB; Class M12, AN 1996-073089 XP002206054 & JP 07 328563 A (OLYMPUS OPTICAL CO LTD), 19 December 1995 (1995-12-19) abstract	10,11
X .	DATABASE WPI Section Ch, Week 199416 Derwent Publications Ltd., London, GB;	10,11
्राध्य transa. ered	Class L03, AN 1994-132722 XP002206055 & JP 06 032795 B (JAPAN FIELD KK), 2 May 1994 (1994-05-02) abstract	
X	DATABASE WPI Section Ch, Week 199439 Derwent Publications Ltd., London, GB; Class A97, AN 1994-312982 XP002206056	10,11
	& JP 06 238243 A (JAPAN FIELD KK), 30 August 1994-(1994-08-30) abstract	
A	US 5 833 761 A (IMAJO YASUTAKA ET AL) 10 November 1998 (1998-11-10) claims 1-12	1-19
A	US 5 676 705 A (RESCH CAROL ET AL) 14 October 1997 (1997-10-14) claims 1-10	1-19
P,A	US 2002/004953 A1 (PERRY ROBERT J ET AL) 17 January 2002 (2002-01-17) claims 1-26	1-19
	-/	

It onal Application No Ful/US 02/04620

	etion) DOCUMENTS CONSIDERED TO BE RELEVANT	PC1703 02	
Category *	Citation of document, with indication, where appropriate, of the relevant passages		Relevant to claim No.
A	US 4 685 930 A (KASPRZAK KENNETH A) 11 August 1987 (1987-08-11) cited in the application claims 1-9		1-19
A	US 5 942 007 A (BERNDT DIETER R ET AL) 24 August 1999 (1999-08-24) cited in the application claims 1-24	·	1-19
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		7 - 4	
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Ir ional Application No

IP IS	53056203 5834416 9910587	A	Publication date 22-05-1978 10-11-1998 04-03-1999	JP JP NONE AU CA EP	Patent family member(s) 1157294 57047960 9038998 2302527	В 	Publication date 15-07-1983 13-10-1982 16-03-1999	
IS	5834416	A	10-11-1998	NONE AU CA EP	9038998 2302527	В 	13-10-1982	· ·
				AU CA EP	2302527			e e
10	9910587	A	04-03-1999	CA EP	2302527			
		, in		EP		8 4		200
مدودا د	-	. 171					04-03-1999	
مدينا د		. *			1007780		14-06-2000	
نديقا د				JP	2001514339		11-09-2001	
				WO	9910587		04-03-1999	
				US US	6010542 6001133		04-01-2000	
				บร	6228826		14-12-1999 08-05-2001	:
D	1000002		10 04 0001			~		
Γ	1092803		18-04-2001					
								:
						WT.	28-02-2002	:
P	6238244	A	30-08-1994	NONE				
P	7328563	Α .	19-12-1995	NONE				
P	6032795	В	02-05-1994	JP	4094779	A	26-03-1992	
Р	6238243	A	30-08-1994	NONE				
S	5833761	A	10-11-1998	US	5503681	Α	02-04-1996	
				AU			24-10-1996	•
	·							
	-3-		•.					
			•	DE				
			,	DE			29-06-2000	
							11-03-1992	* 4
							19-06-1996	
								٠.
			1					-
٠	:							
				JP				
				KR	145061	B1	17-08-1998	
				KR	9510446	B1	18-09-1995	٠.
							15-11-1997	٠.
								,
								•
.•				US				
	P P P	P 1092803 P 6238244 P 7328563 P 6032795 P 6238243 S 5833761	P 6238244 A P 7328563 A P 6032795 B P 6238243 A	P 6238244 A 30-08-1994 P 7328563 A 19-12-1995 P 6032795 B 02-05-1994 P 6238243 A 30-08-1994 S 5833761 A 10-11-1998	EP US P 6238244 A 30-08-1994 NONE P 7328563 A 19-12-1995 NONE P 6032795 B 02-05-1994 JP P 6238243 A 30-08-1994 NONE S 5833761 A 10-11-1998 US AU AU AU CA DE DE DE EP EP EP EP EP EP EP	EP 1092803 US 2002023305 P 6238244 A 30-08-1994 NONE P 7328563 A 19-12-1995 NONE P 6032795 B 02-05-1994 JP 4094779 P 6238243 A 30-08-1994 NONE S 5833761 A 10-11-1998 US 5503681 AU 6314594 AU 646246 AU 7475191 CA 2050333 DE 69126648 DE 69131853 DE 69131853 DE 69131853 DE 69131853 DE 69131853 US 5763270 JP 7275812 JP 3115781 JP 7275813 WO 9113697 JP 2723359 KR 145061 KR 9510446 KR 9701233 SG 50628 SG 44844 SG 48276 US 5888312 US 5769962 US 5772781	EP 1092803 A1 US 2002023305 A1 P 6238244 A 30-08-1994 NONE P 7328563 A 19-12-1995 NONE P 6032795 B 02-05-1994 JP 4094779 A P 6238243 A 30-08-1994 NONE S 5833761 A 10-11-1998 US 5503681 A AU 673062 B2 AU 6314594 A AU 646246 B2 AU 7475191 A CA 2050333 A1 DE 69126648 D1 DE 69126648 D1 DE 69126648 D1 DE 69131853 D1 DE 69131853 T2 EP 0473795 A1 EP 0717128 A2 EP 0717129 A2 EP 0757119 A2 JP 7275812 A JP 3115781 B2 JP 7275812 A JP 3115781 B2 JP 7275812 A JP 3115781 B2 JP 7275813 A WO 9113697 A1 JP 2723359 B2 KR 145061 B1 KR 9510446 B1 KR 9701233 B1 SG 50628 A1 SG 48276 A1 US 5888312 A US 5769962 A US 5769962 A US 5772781 A	EP 1092803 A1 18-04-2001 US 2002023305 A1 28-02-2002 P 6238244 A 30-08-1994 NONE P 7328563 A 19-12-1995 NONE P 6032795 B 02-05-1994 JP 4094779 A 26-03-1992 P 6238243 A 30-08-1994 NONE S 5833761 A 10-11-1998 US 5503681 A 02-04-1996 AU 673062 B2 24-10-1996 AU 673062 B2 24-10-1996 AU 673062 B2 17-02-1994 AU 646246 B2 17-02-1994 AU 7475191 A 10-10-1991 CA 2050333 A1 17-09-1991 DE 69126648 D1 31-07-1997 DE 69131853 D1 20-01-2000 DE 69131853 D1 20-01-2000 DE 69131853 T2 29-06-2000 DF 0473795 A1 11-03-1992 DF 0473795 A1 11-03-1992 DF 0757119 A2 19-06-1996 DF 0757119 A2 19-06-1998 DF 0757119 A2 19-09-1991 DF 0725812 A 24-10-1995 DF 0757119 A2 19-09-1991 DF 0725812 A 24-10-1995 DF 0757119 A2 19-06-1998 DF 0757119 A2 19-09-1991 DF 075812 A 24-10-1995 DF 0757119 A2 19-06-1998 DF 0757119 A2 19-06-1998 DF 0757119 A2 19-06-1998 DF 0757119 A2 19-06-1999 DF 0757119 A2 19-06-1999 DF 0757119 A2 19-06-1999 DF 0757119 A2 19-06-1998 DF 0757119 A2 19-06-1999 DF 0757119 A2 19-06-1998 DF 0757119 A2 19-06-1990

nformation on patent family members

Ir donal Application No

			101,00	02/ 04020
Patent document cited in search report	Publication date		Patent family member(s)	Publication date
US 5676705 A	14-10-1997	AU	4942996 A	23-09-1996
1	11 10 100,	CA	2211412 A1	12-09-1996
	9	WO	9627704 A1	12-09-1996
		EP,	0813628 A1	29-12-1997
		FI	973603 A	05-09-1997
·		TR	9700901 T1	21-02-1998
		ÜS	5683473 A	04-11-1997
		ZA	9601786 A	05-09-1997
	·			
US 2002004953 A1	17-01-2002	WO	0250366 A1	27-06-2002
US 4685930 A	11-08-1987	AU	585906 B2	29-06-1989
		AU	4981785 A	22-05-1986
		CA	1239326 A1	19-07-1988
		DE.	3583377 01	08-08-1991
		EP	0182583 A2	28-05-1986
		JP	1502875 C	28-06-1989
		JP	61119765 A	06-06-1986
		JP	63050463 B	07-10-1988
US 5942007 A	24-08-1999	US	5865852 A	02-02-1999
		AT	215631 T	15-04-2002
		AU	4993599 A	07-02-2000
		AU .	5101799 A	07-02-2000
	v	BR	9912088 A	10-04-2001
		BR	9912811 A	02-05-2001
	·	CN	1309733 T	22-08-2001
· · · · · · · · · · · · · · · · · · ·		CN	1309734 T	22-08-2001
		DE	69901179 D1	08-05-2002
		DK	1084289 T3	24-06-2002
		EP	1084289 A1	21-03-2001 18-04-2001
		EP Hu	1092056 A1 0102648 A2	28-12-2001
		NO	20010231 A	14-03-2001
	•	NO	20010231 A 20010232 A	14-03-2001
		PL	345504 A1	17-12-2001
	•	PL .	345509 A1	17-12-2001
		TR	200100312 T2	23-07-2001
]		TR	200100312 T2 200100318 T2	21-08-2001
		WO	0004221 A1	27-01-2000
		WO	0004222 A1	27-01-2000
· ·		US	6063135 A	16-05-2000
		US	6042617 A	28-03-2000
	*	US	6056789 A	02-05-2000
1.		US	6042618 A	28-03-2000
		US	6086635 A	11-07-2000
·		US	6059845 A	09-05-2000
· · · · · · · · · · · · · · · · · · ·	·		,	

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